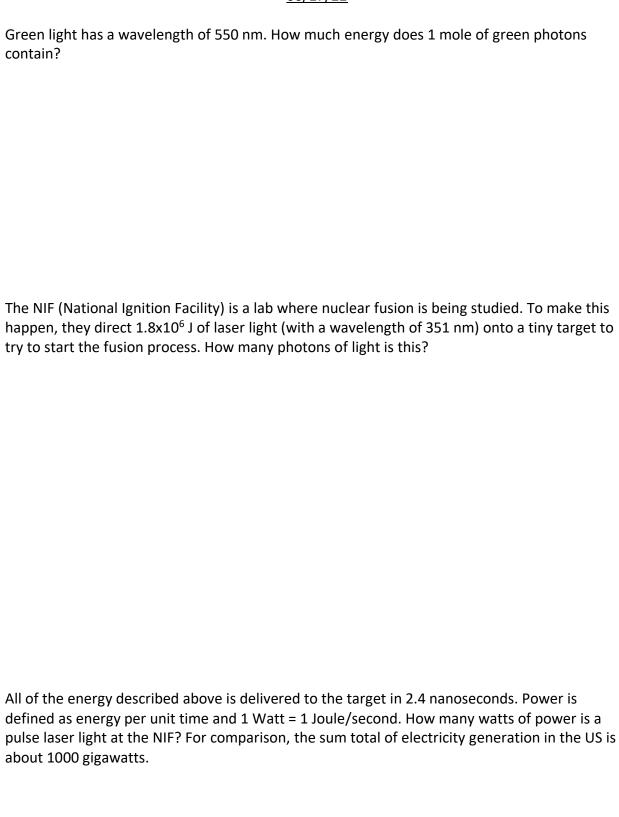
06/17/22



If you want to excite an electron from the n=3 level to the n=6 level, you can shine light of the appropriate wavelength/frequency on the sample. What wavelength of light is required to do this?	that ends at the naqua line is 486.1	rom the hydrogen dischane level. The wavelengthom and the wavelengthoe each line (i.e. which niles	of the red line is 656. of the blue/purple line	2 nm, the wavelength is 434.0 nm. Which t	n of the
appropriate wavelength/frequency on the sample. What wavelength of light is required to do					
appropriate wavelength/frequency on the sample. What wavelength of light is required to do					
appropriate wavelength/frequency on the sample. What wavelength of light is required to do	If you want to exc	cite an electron from the	n=3 level to the n=6 le	vel, you can shine lig	ht of the
	appropriate wave			-	