

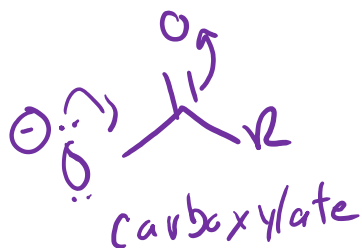
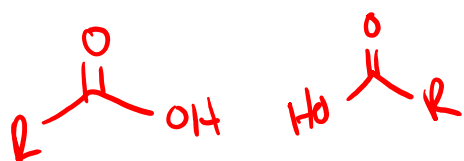
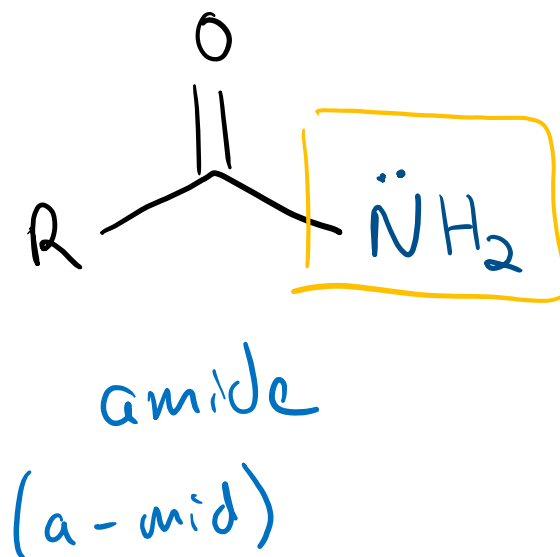
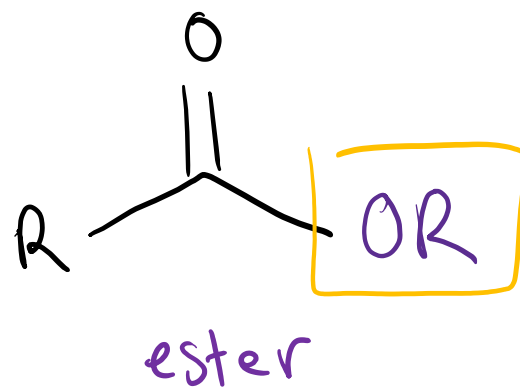
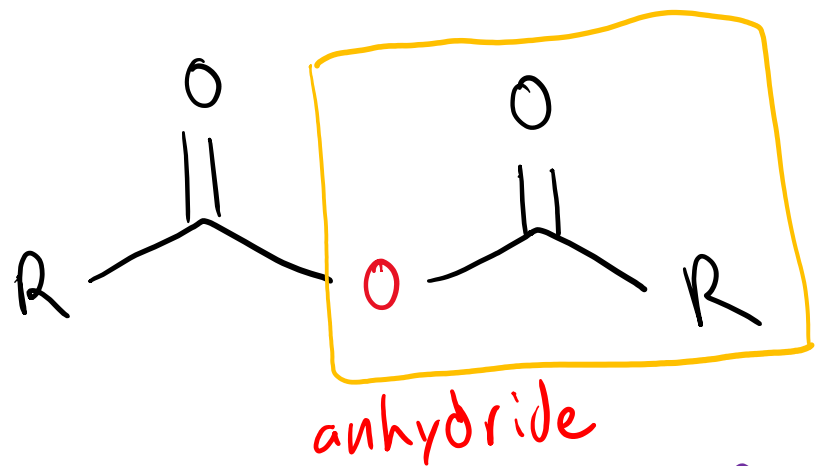
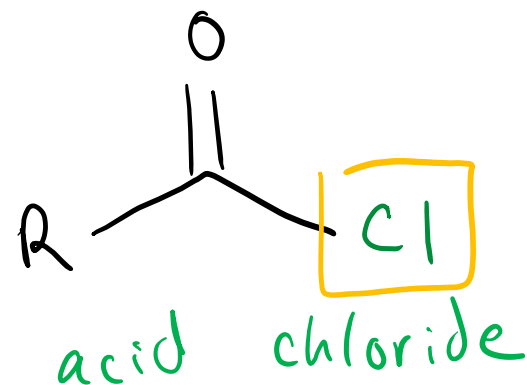
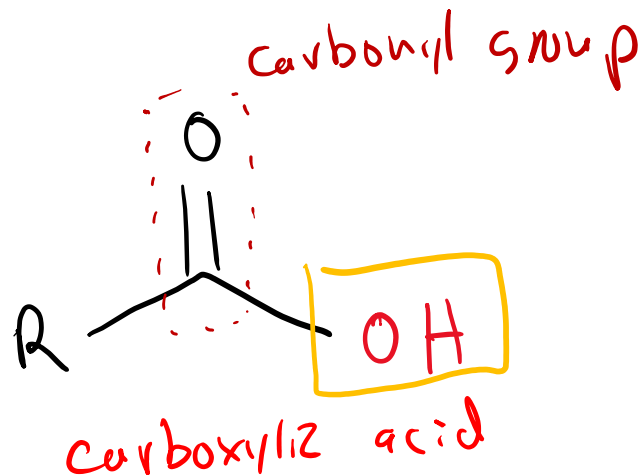
Carboxylic Acids & Their Derivatives

2/20/2023

Ch. 20: Carboxylic Acids and Their Derivatives

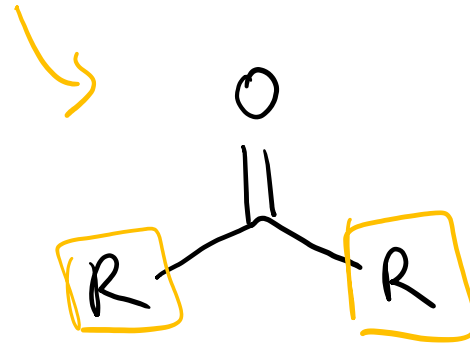
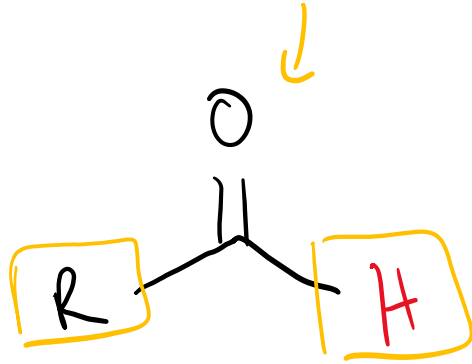
(20.6)

boxed group
= leaving group
(may require
protonation)

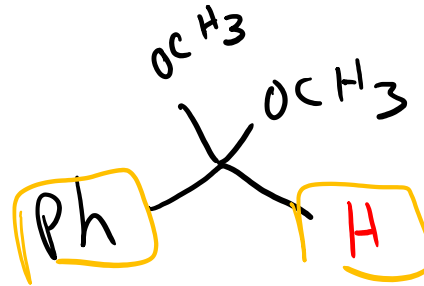
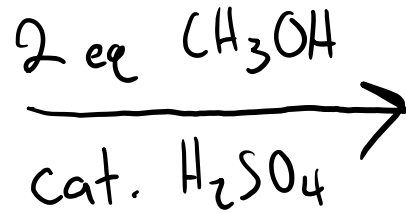
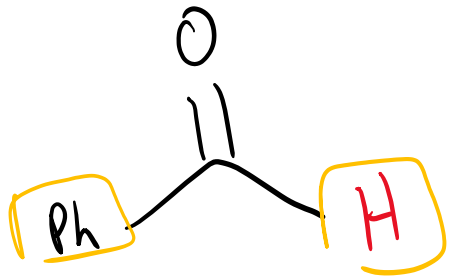


Comparison with Aldehyde and Ketone Reactivity

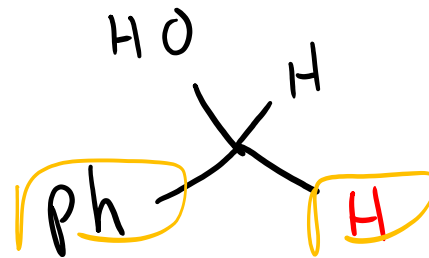
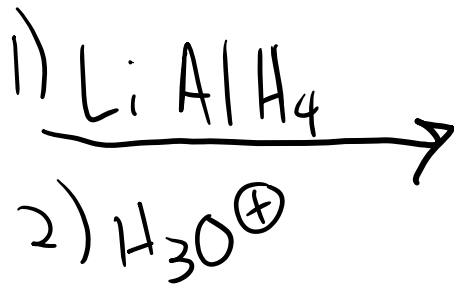
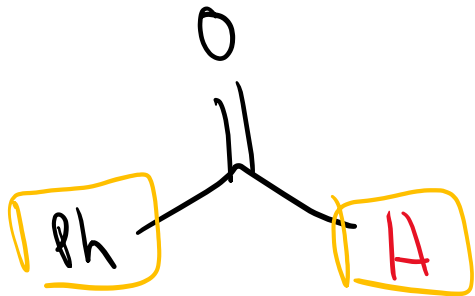
$\text{H}:\ominus$ bad LG!
 $\text{pK}_a \text{H} (\text{H}_2) \sim 35$



$\text{R}:\ominus$ bad LG!
 $\text{pK}_a \text{H} (\text{alkane}) = 50$

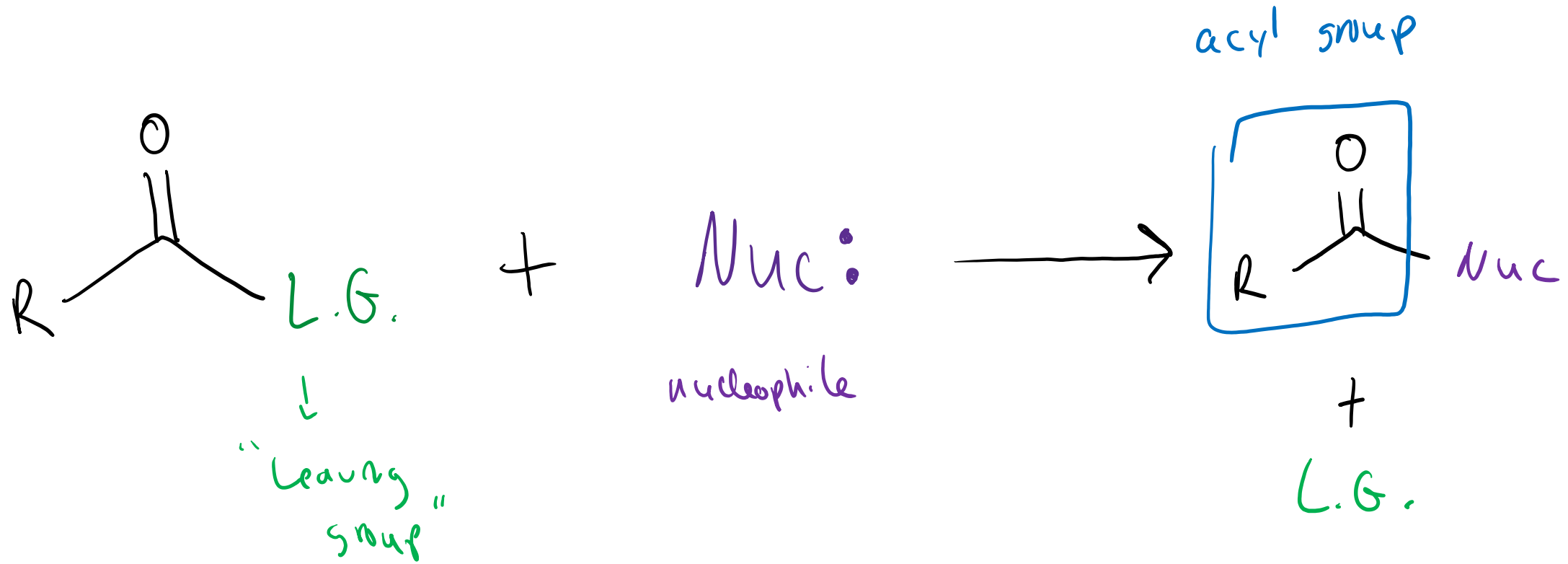


stayed intact.
did not leave!



no $\text{C}=\text{O}$ in product!

Acyl Transfer Reaction (general)



"looks like" S_N2 , but technically not.

bc goes through tetrahedral intermediate.

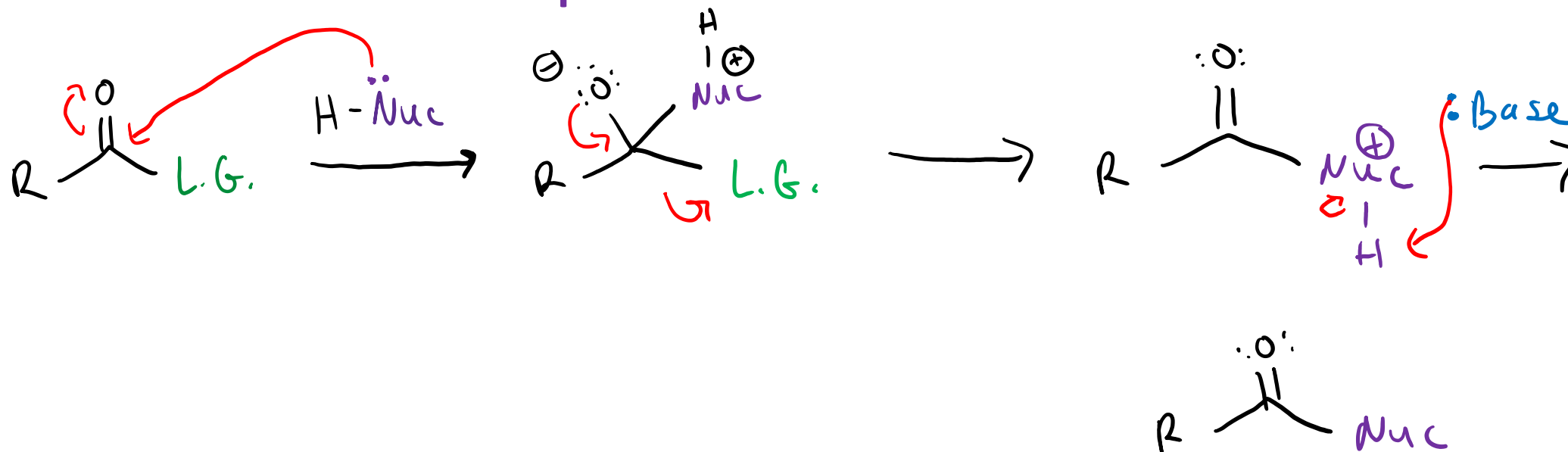
Acyl Transfer Mechanisms (General)

(20.7)

With an **anionic nucleophile**: (for derivatives w/ best leaving groups)



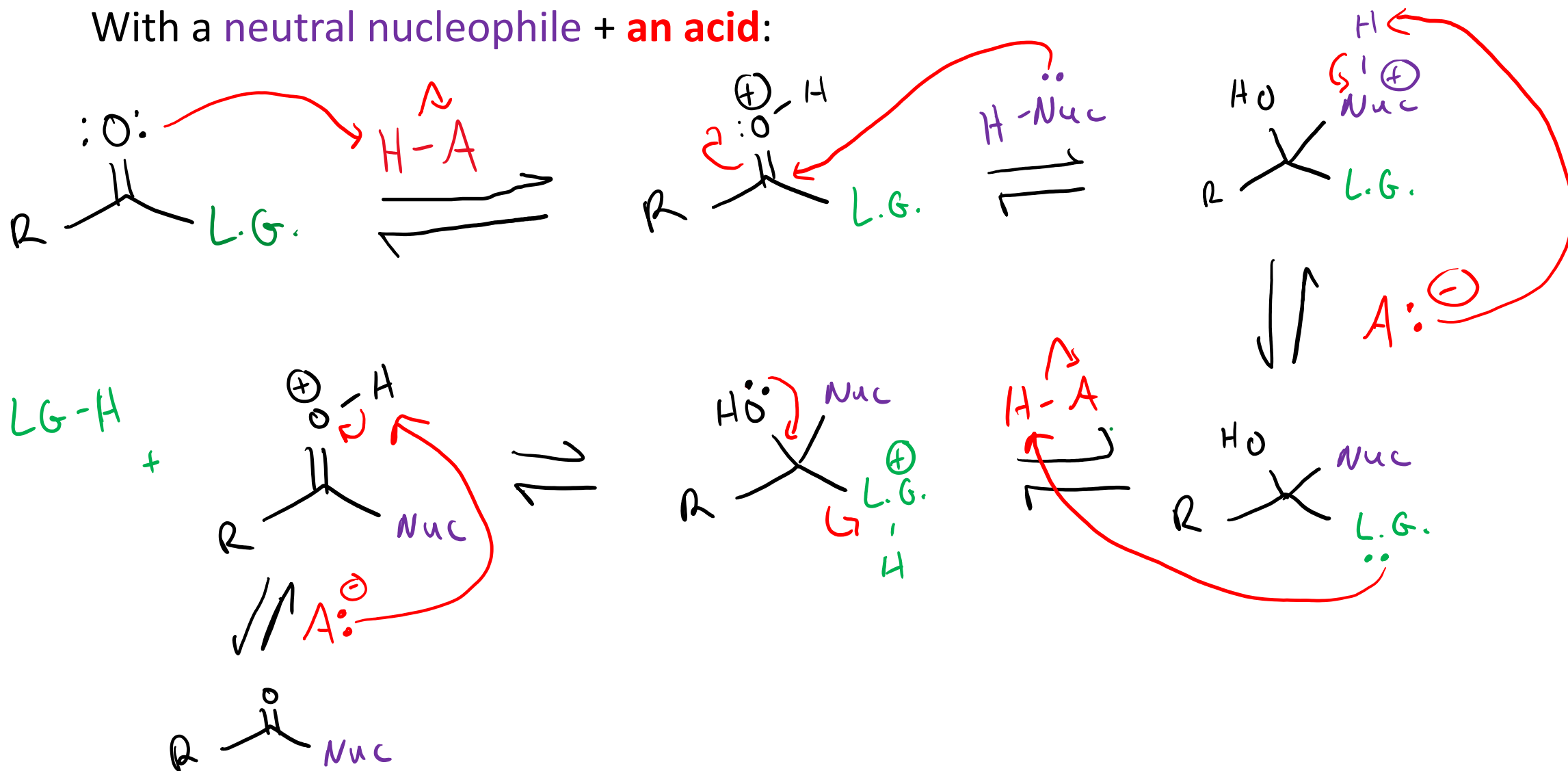
With a **neutral nucleophile** + a **weak base**



Acyl Transfer Mechanisms (General)

(20.7)

With a **neutral nucleophile** + **an acid**:



Ranking Electrophilicity of Carbonyl Functional Groups

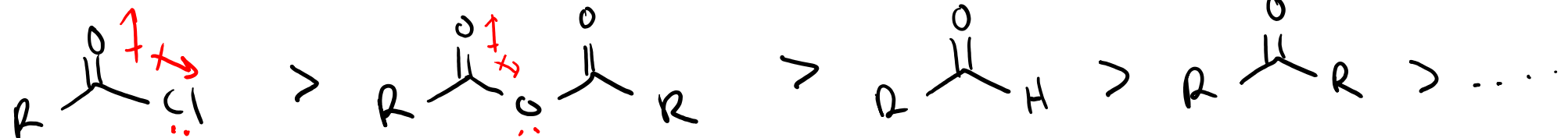
(20.7)

acid chloride

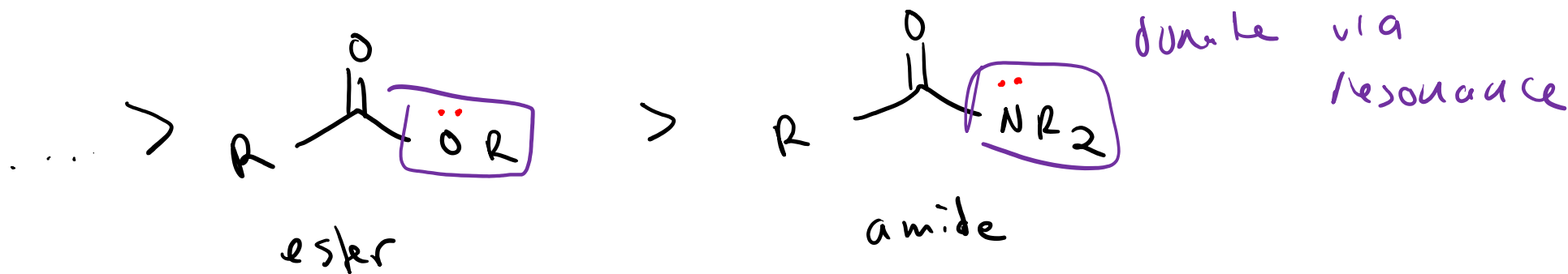
anhydride

aldehyde

ketone



most
 E^+



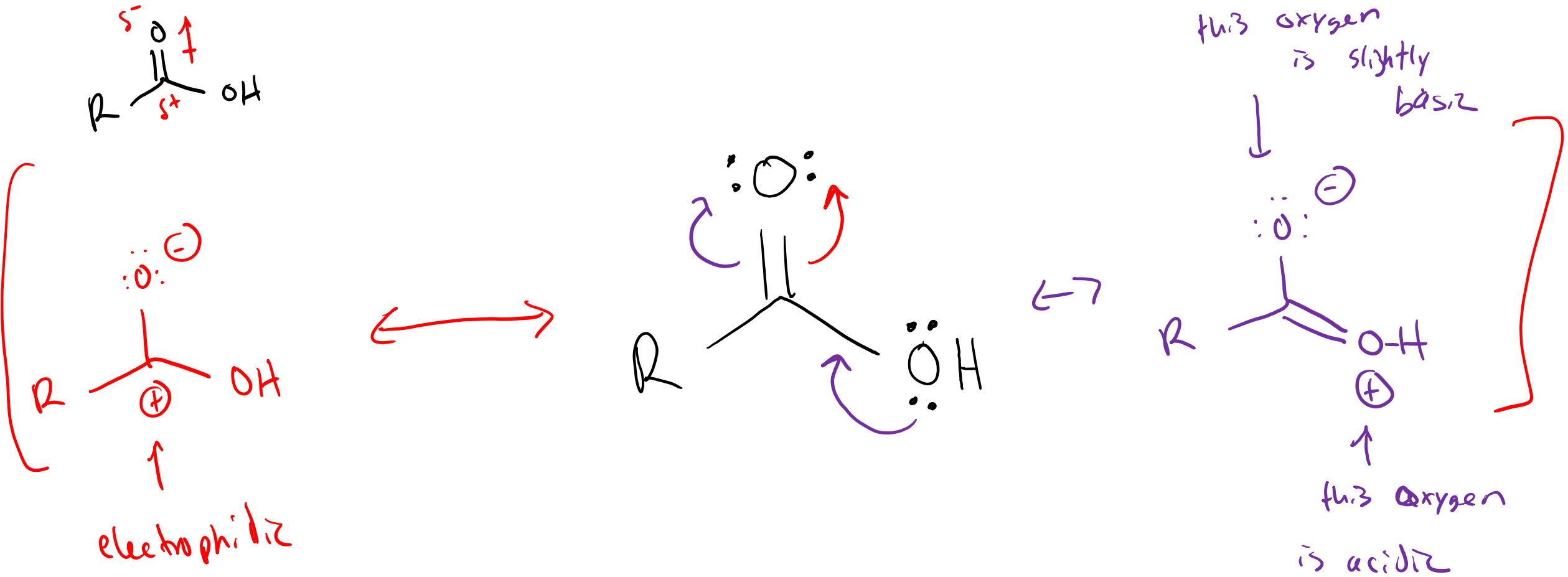
don't be via
resonance

least

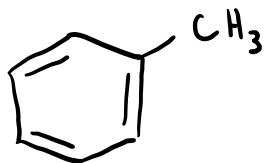
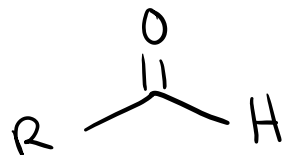
E^+

Carboxylic Acids

(20.3)



Synthesis of Carboxylic Acids



★ only new one ★
hydrolysis of nitriles!

