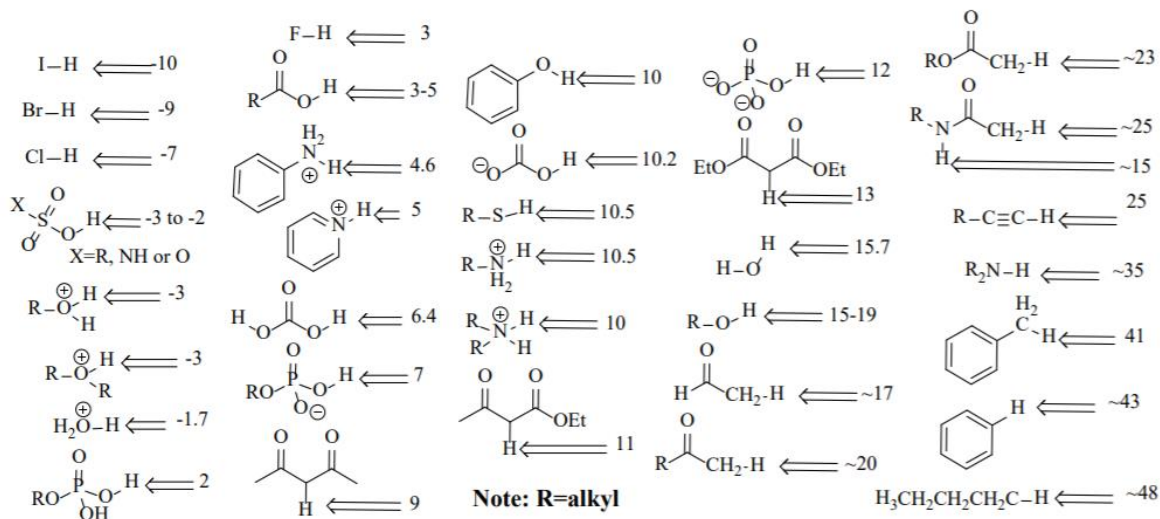


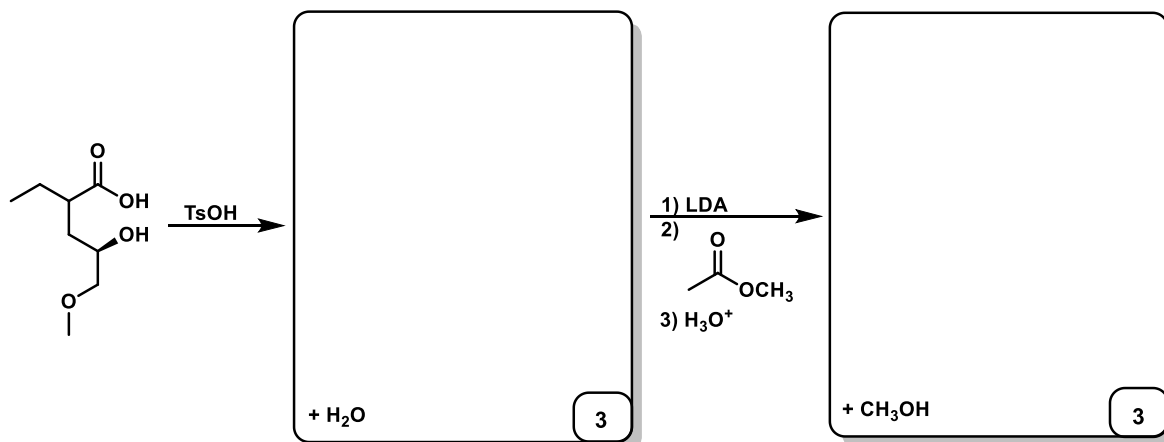
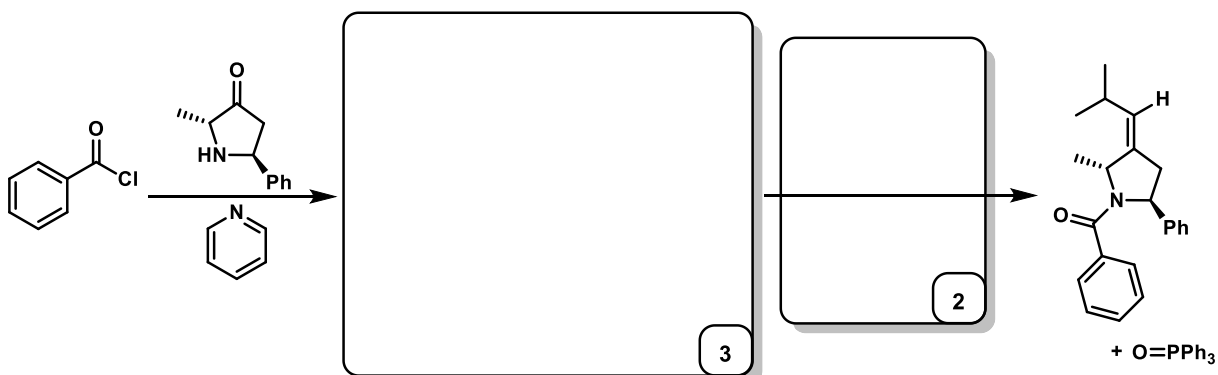
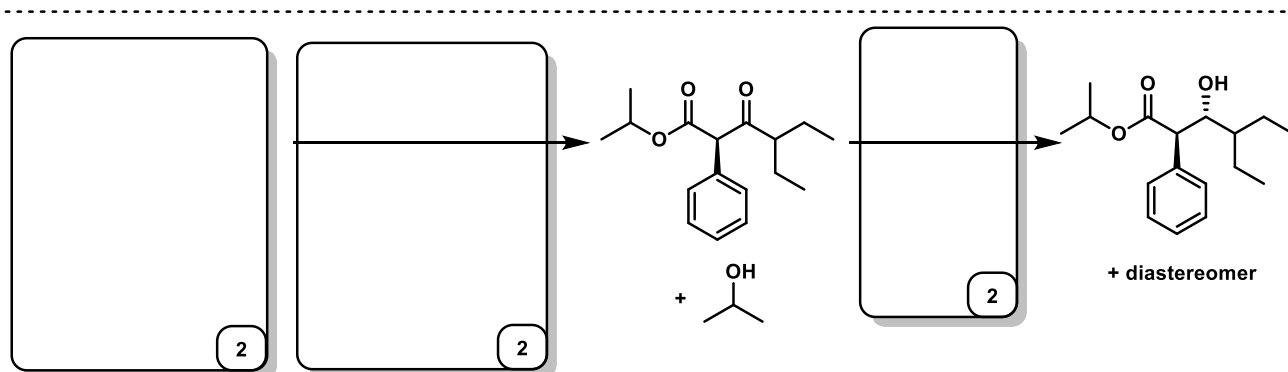
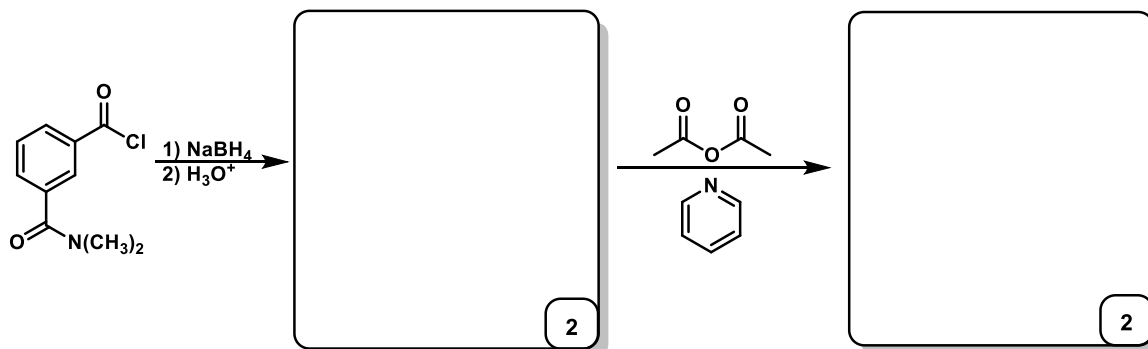
Name: \_\_\_\_\_

| IA                |  |                     |  |                      |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  | 0                |  |
|-------------------|--|---------------------|--|----------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|-------------------|--|------------------|--|
| 1<br>H<br>1.008   |  | IIA                 |  |                      |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  | 2<br>He<br>4.003 |  |
| 3<br>Li<br>6.941  |  | 4<br>Be<br>9.012    |  |                      |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  | 5<br>B<br>10.81   |  | 6<br>C<br>12.01   |  | 7<br>N<br>14.01   |  | 8<br>O<br>16.00   |  | 9<br>F<br>19.00   |  | 10<br>Ne<br>20.18 |  |                   |  |                   |  |                  |  |
| 11<br>Na<br>22.99 |  | 12<br>Mg<br>24.31   |  | IIIB                 |  | IVB               |  | VB                |  | VIB               |  | VIIB              |  | VIIIB             |  |                   |  |                   |  | IB                |  | IIB               |  | 13<br>Al<br>26.98 |  | 14<br>Si<br>28.09 |  | 15<br>P<br>30.97  |  | 16<br>S<br>32.06  |  | 17<br>Cl<br>35.45 |  | 18<br>Ar<br>39.95 |  |                  |  |
| 19<br>K<br>39.10  |  | 20<br>Ca<br>40.08   |  | 21<br>Sc<br>44.96    |  | 22<br>Ti<br>47.90 |  | 23<br>V<br>50.94  |  | 24<br>Cr<br>52.00 |  | 25<br>Mn<br>54.94 |  | 26<br>Fe<br>55.85 |  | 27<br>Co<br>58.93 |  | 28<br>Ni<br>58.70 |  | 29<br>Cu<br>63.55 |  | 30<br>Zn<br>65.38 |  | 31<br>Ga<br>69.72 |  | 32<br>Ge<br>72.59 |  | 33<br>As<br>74.92 |  | 34<br>Se<br>78.96 |  | 35<br>Br<br>79.90 |  | 36<br>Kr<br>83.80 |  |                  |  |
| 37<br>Rb<br>85.47 |  | 38<br>Sr<br>87.62   |  | 39<br>Y<br>88.91     |  | 40<br>Zr<br>91.22 |  | 41<br>Nb<br>92.91 |  | 42<br>Mo<br>95.94 |  | 43<br>Tc<br>(98)  |  | 44<br>Ru<br>101.1 |  | 45<br>Rh<br>102.9 |  | 46<br>Pd<br>106.4 |  | 47<br>Ag<br>107.9 |  | 48<br>Cd<br>112.4 |  | 49<br>In<br>114.8 |  | 50<br>Sn<br>118.7 |  | 51<br>Sb<br>121.8 |  | 52<br>Te<br>127.6 |  | 53<br>I<br>126.9  |  | 54<br>Xe<br>131.3 |  |                  |  |
| 55<br>Cs<br>132.9 |  | 56<br>Ba<br>137.3   |  | 57 *<br>La<br>138.9  |  | 72<br>Hf<br>178.5 |  | 73<br>Ta<br>180.9 |  | 74<br>W<br>183.9  |  | 75<br>Re<br>186.2 |  | 76<br>Os<br>190.2 |  | 77<br>Ir<br>192.2 |  | 78<br>Pt<br>195.1 |  | 79<br>Au<br>197.0 |  | 80<br>Hg<br>200.6 |  | 81<br>Tl<br>204.4 |  | 82<br>Pb<br>207.2 |  | 83<br>Bi<br>209.0 |  | 84<br>Po<br>(209) |  | 85<br>At<br>(210) |  | 86<br>Rn<br>(222) |  |                  |  |
| 87<br>Fr<br>(223) |  | 88<br>Ra<br>(226.0) |  | 89 **<br>Ac<br>(227) |  | 104<br>Rf         |  | 105<br>Ha         |  | 106<br>Unh        |  | 107<br>Uns        |  | 108               |  | 109<br>Une        |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                   |  |                  |  |

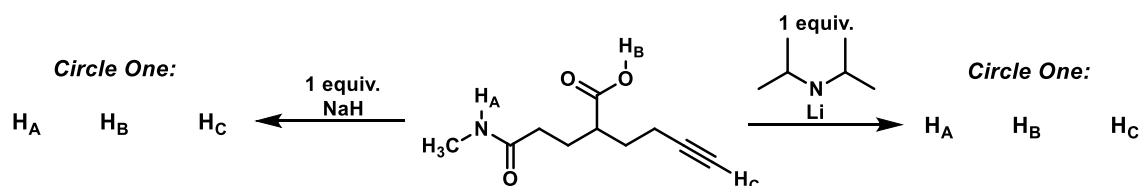
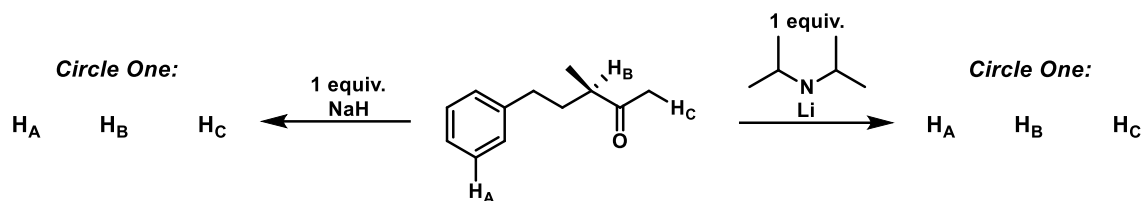
|                   |                   |                   |                   |                   |                   |                   |                   |                   |                   |                    |                    |                    |                    |
|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|--------------------|--------------------|--------------------|--------------------|
| 58<br>Ce<br>140.1 | 59<br>Pr<br>140.9 | 60<br>Nd<br>144.2 | 61<br>Pm<br>(145) | 62<br>Sm<br>150.4 | 63<br>Eu<br>152.0 | 64<br>Gd<br>157.3 | 65<br>Tb<br>158.9 | 66<br>Dy<br>162.5 | 67<br>Ho<br>164.9 | 68<br>Er<br>167.3  | 69<br>Tm<br>168.9  | 70<br>Yb<br>173.0  | 71<br>Lu<br>175.0  |
| 90<br>Th<br>232.0 | 91<br>Pa<br>(231) | 92<br>U<br>238.0  | 93<br>Np<br>(244) | 94<br>Pu<br>(242) | 95<br>Am<br>(243) | 96<br>Cm<br>(247) | 97<br>Bk<br>(247) | 98<br>Cf<br>(251) | 99<br>Es<br>(252) | 100<br>Fm<br>(257) | 101<br>Md<br>(258) | 102<br>No<br>(259) | 103<br>Lr<br>(260) |

pK<sub>a</sub> information

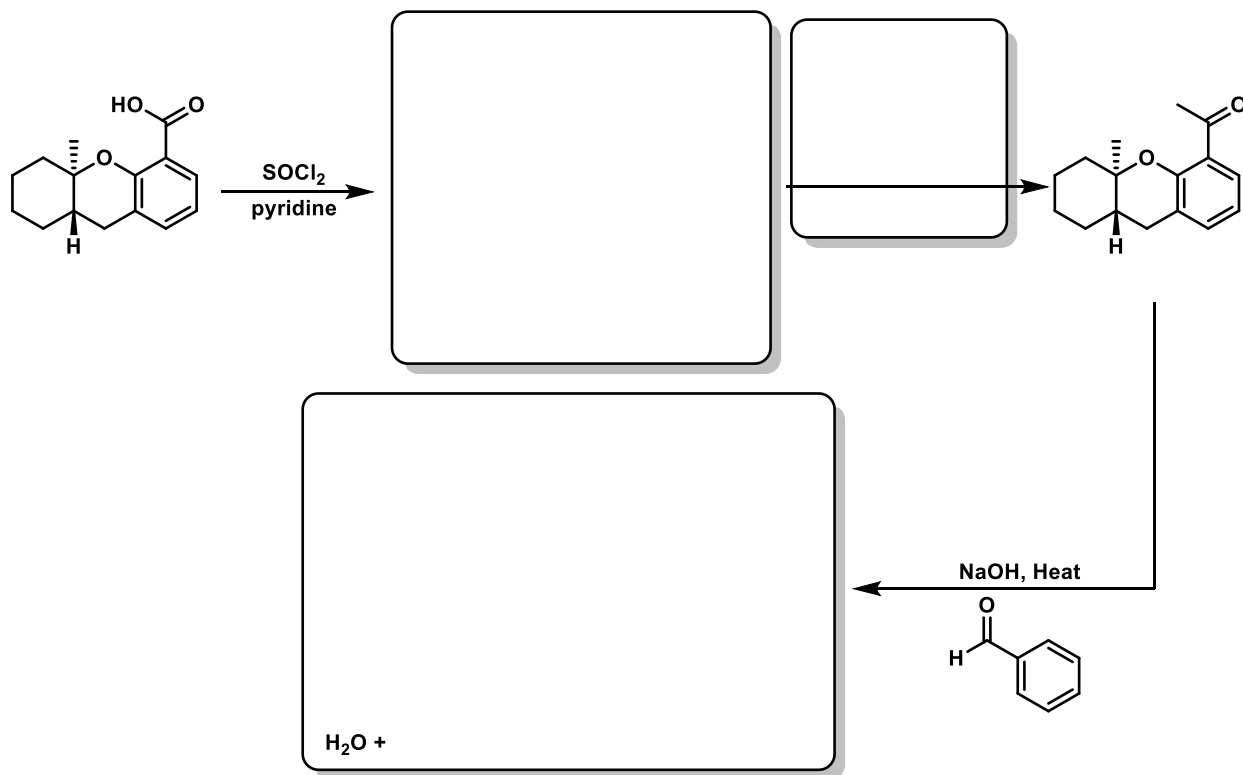
1. **Quick Synthesis.** Fill in the boxes with either the correct starting materials, reagents or products. You can ignore stereochemistry for all enolate-derived chemistry, but you should consider stereochemistry for all other chemistry, where appropriate. (21 points)



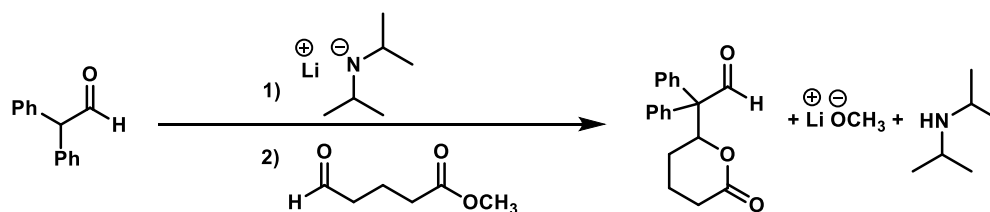
2. **Acid / Base.** For each acid / base reaction, circle the correct hydrogen that would be deprotonated under the conditions indicated. (8 pts)



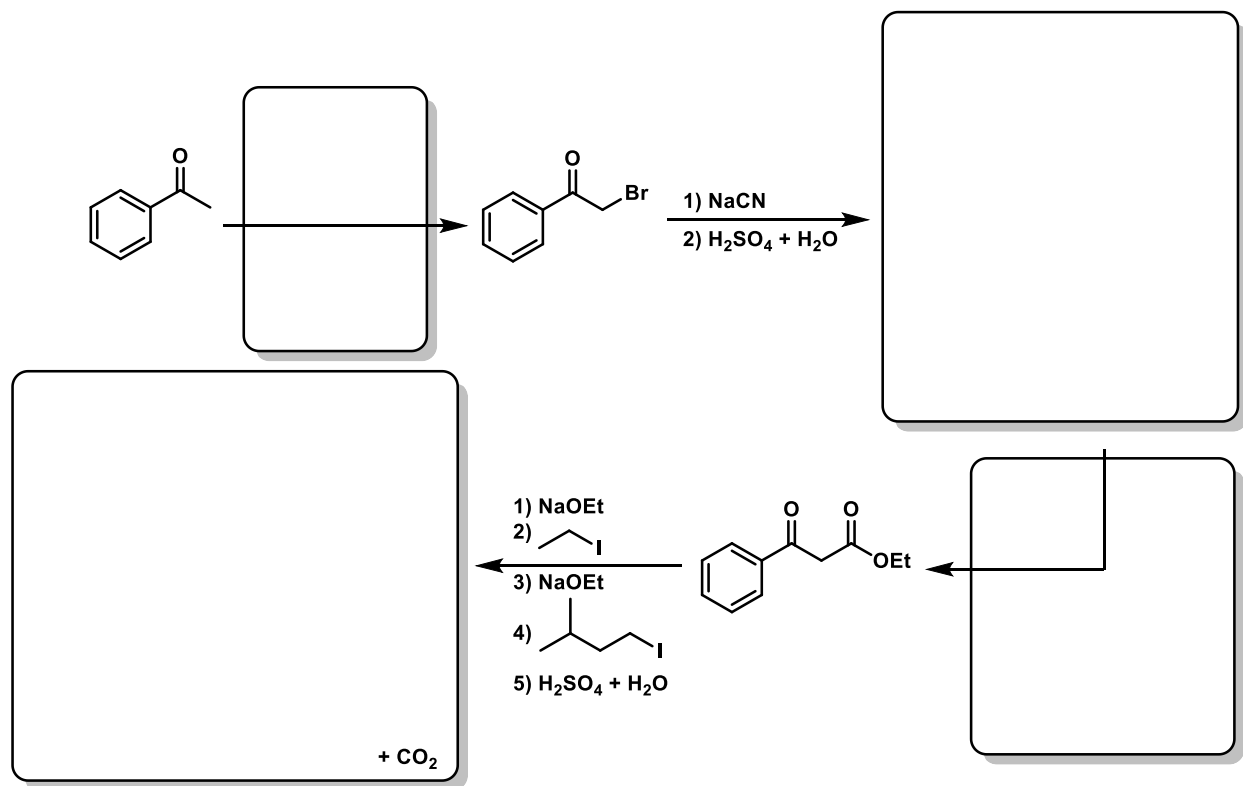
5. **Guided Synthesis.** Fill in the missing reagents / intermediates / products to complete the following synthetic scheme. (12 pts)



3. **Mechanism.** Provide a step-wise mechanism for the following transformation. *Hint:* the correct mechanism will combine chemistry / mechanisms you learned in both Ch 20 and Ch 21. (15 points)

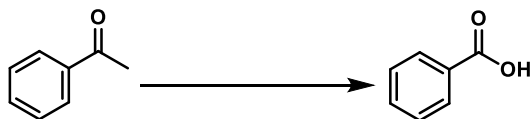


6. **Guided Synthesis.** Fill in the missing reagents / intermediates / products to complete the following synthetic schemes. (12 points)



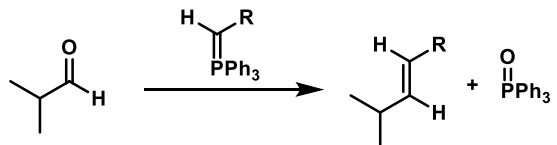
7. **Multiple Choice.** Select the letter that best answers the questions posed. (10 points)

Which of the following sets of conditions would produce the desired product?



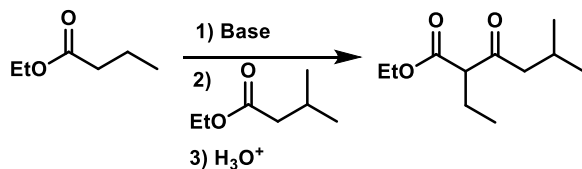
- A. 1)  $\text{Br}_2 + \text{NaOH}$  2)  $\text{H}_3\text{O}^+$     B. 1) NaOH 2)  $\text{H}_3\text{O}^+$     C. 1)  $\text{H}_3\text{CMgBr}$  2)  $\text{H}_3\text{O}^+$

Which ylide would provide the stereochemical outcome shown below?



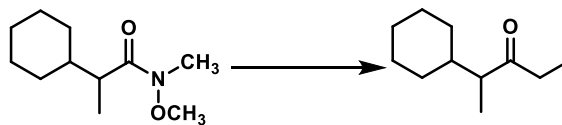
- A. H-C(=P(Ph)3)-C1CCCCC1    B. H-C(=P(Ph)3)-C(=O)OC    C. H-C(=P(Ph)3)-c1ccccc1

What is the best base choice to complete the following transformation?



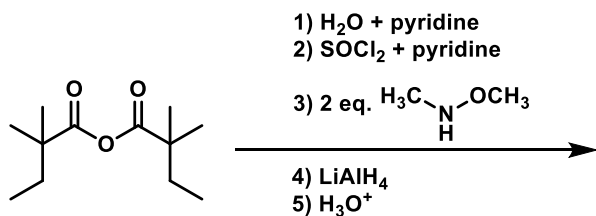
- A. NaOEt    B. LDA    C.  $\text{NaOCH}_3$

Which reagents would accomplish this reaction?



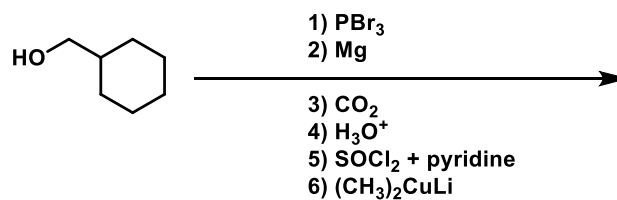
- A. 1)  $\text{LiAlH}_4$  2)  $\text{H}_3\text{O}^+$     B. 1) CCMgBr 2)  $\text{H}_3\text{O}^+$     C. 1) CCLi 2)  $\text{H}_3\text{O}^+$

8. **Less-Guided Synthesis.** First, complete the first two synthetic pathways to determine *Product 6A* and *Product 6B*. Then, in the final box, combine these two products in the manner described to get the final product of the synthesis. (12 points)



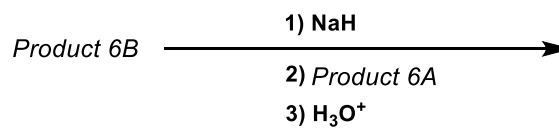
*Product 6A*

5



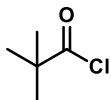
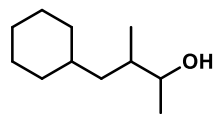
*Product 6B*

5

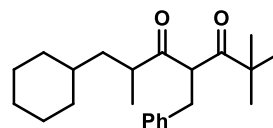


2

9. **Full Synthesis.** The **Desired Product** below can be made from the given **Starting Materials** using reactions we have learned in class. You must use both of the starting materials. Provide a synthesis in the box below. *Hint:* you'll want to alter both of the starting materials before reacting them with each other. (10 points)



**Starting Materials**



**Desired Product**