

7/12 & 7/15 Eq. recap

- Determine K_c or K_p & whether the reaction is product or reactant favored
① eq.
- Determine the new eq. constant if a reaction is reversed, multiplied, or two reactions add to give the original reaction
- Determine eq. concentrations or eq. pressures if given K and are or more initial concentration.
or pressure

• Apply Le-Chatelier's
to an eq. reaction.

- change in pressure
- change in temp.
- change in $[]$
- change in volume